

1. (b) $\alpha = 3 \times 10^8 \text{ m}$

$$\text{Distance} = (400) \times 3 \times 10^8 \text{ m} = 400\alpha$$

2. (a) As per the question, considering H as magnetic field

$$[H] = \frac{MLT^{-2}}{LT^{-1}IT} = MI^{-1} T^{-2}$$

$$[E] = \frac{MLT^{-2}}{IT} = MLI^{-1} T^{-3}$$

$$[\epsilon] = \frac{I^2 T^2}{MLT^{-2} L^2} = M^{-1} L^{-3} I^2 T^4$$

$$MI^{-1} T^{-2} = M^{-q+r} L^{p-3q+r} I^{2q-r} T^{4q-3r-s}$$

$$-q + r = 1$$

$$p - 3q + r = 0$$

$$2q - r = -1$$

$$4q - 3r - s = -2$$

Solving we get

$$q = 0, r = 1$$

$$p = -1, s = -1$$

No option matching

Considering H as magnetic field intensity

$$H = \frac{B}{\mu_0} = Ni \Rightarrow \text{current per unit length} \Rightarrow IL^{-1}$$

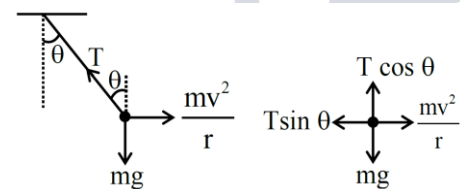
$$-q + r = 0 \quad q = 1$$

$$p - 3q + r = -1 \quad r = 1$$

$$2q - r = 1 \quad p = +1$$

$$4q - 3r - s = 0 \quad s = +1$$

3. (c)



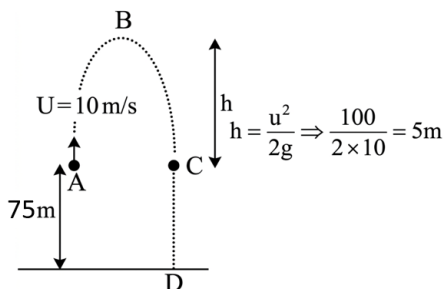
$$\tan \theta = \frac{v^2}{rg} \left(v = 54 \times \frac{5}{18} = 15 \text{ m/s} \right)$$

$$\tan \theta = \frac{225}{20 \times 10}$$

$$\tan \theta = \frac{9}{8}$$

$$\theta = \tan^{-1}(1.125)$$

4. (d)



$$h = \frac{u^2}{2g} \Rightarrow \frac{100}{2 \times 10} = 5 \text{ m}$$

$$-75 = 10 \times t - \frac{1}{2} \times 10 \times t^2$$

$$t^2 - 2t - 15 = 0$$

$$(t - 5)(t + 3) = 0 \Rightarrow t = 5$$

$$\text{Height} = 75 + 10 \times 5 = 125 \text{ m}$$

5. (a) Combination of lens + mirror

$$f_m = \frac{R}{2} = -10 \text{ cm}; f_L = \frac{R}{2(\mu-1)} = 20 \text{ cm}$$

$$\frac{1}{f} = \frac{1}{f_m} - \frac{2}{f_L}$$

$$\frac{1}{f} = \frac{1}{-10} - \frac{2}{20} = -\frac{1}{5}$$

$$f = -5 \text{ cm}$$

To form image at the position of object, it should be placed at center of curvature of mirror + lens combination.

$$\therefore u = 2f = 2 \times 5 = 10 \text{ cm}$$

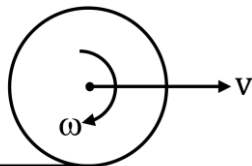
6. (a) For range to be same, angle of projection must be complementary angles.

$$T_1 = \frac{2u \sin \theta}{g} = 10, T_2 = \frac{2u \cos \theta}{g} = 5$$

$$\text{Range} = \frac{2(u \sin \theta)(u \cos \theta)}{g}$$

$$= \frac{2(50)(25)}{10} \text{ m} = 250 \text{ m}$$

7. (b)



$$a = \frac{\mu_K mg}{m} = \mu_K g$$

$$\tau = I \alpha \Rightarrow \mu_K mg R = \frac{MR^2}{2} \times \alpha$$

$$\Rightarrow \alpha = \frac{2\mu_K g}{R}$$

$$\therefore v = v_0 - \mu_K g t \quad \dots(i)$$

$$\text{Also, } \omega = \omega_0 + \frac{2\mu_K g t}{R} \quad \dots(ii)$$

$$\frac{v}{R} = \frac{v_0}{4R} + \frac{2\mu_K g t}{R}$$

Solving Eq. (i) & (ii)

$$\frac{v_0}{4} + 2\mu_K g t = v_0 - \mu_K g t$$

$$3\mu_K g t = v_0 - \frac{v_0}{4}$$

$$3\mu_K g t = \frac{3v_0}{4}$$

$$\Rightarrow t = \frac{v_0}{4\mu_K g} = \frac{49}{4 \times \frac{1}{4} \times 9.8} = 5 \text{ sec.}$$

8. (d) Using continuity equation,

$$A_1 V_1 = A_2 V_2$$

$$1 \times 10 = \frac{1}{5} \times V_2$$

$$\Rightarrow V_2 = 50 \text{ cm/s}$$

Using Bernoulli's theorem,

$$P_1 + \frac{1}{2} \rho V_1^2 = P_2 + \frac{1}{2} \rho V_2^2$$

$$P_1 - P_2 = \frac{1}{2} \rho [V_2^2 - V_1^2]$$

$$= \frac{1}{2} \times 600 [(50)^2 - (10)^2] \times 10^{-4} = 72 \text{ Pa.}$$

9. (d) Using volume conservation,

$$\frac{4}{3}\pi R_1^3 = 64 \times \frac{4}{3}\pi R_2^3 \Rightarrow R_2 = \frac{R_1}{4}$$

Terminal velocity $v \propto R^2$

$$\frac{v_1}{v_2} \Rightarrow \frac{R_1^2}{R_2^2} \Rightarrow (4)^2 = 16$$

10. (b) Method-1

$$Q = \Delta U + W$$

$$Q = \frac{f}{2}nR\Delta T + nR\Delta T = \left(\frac{f}{2} + 1\right)nR\Delta T$$

$$\text{if } f = 2 \Rightarrow Q = 2nR\Delta T \Rightarrow 2 \times 1 \times 8.3 \times 1.2 = 19.92$$

(No option is matching)

$$\text{if } f = 5 \Rightarrow Q = 3.5nR\Delta T$$

$$\Rightarrow 3.5 \times 1 \times 8.3 \times 1.2 = 34.86$$

(No option is matching)

Method-2

$$Q = \Delta U + W$$

$$Q = \frac{f}{2}nR\Delta T + P\Delta V$$

$$\text{if } f = 2$$

$$\Rightarrow Q = nR\Delta T + P\Delta V$$

$$= 1 \times 8.3 \times 1.2 + 10^5 \times 25 \times 10^{-3} \times 4 \times 10^{-4}$$

$$= 10.96$$

(No option is matching)

$$\text{if } f = 5$$

$$\Rightarrow Q = 2.5nR\Delta T + P\Delta V$$

$$2.5 \times 1 \times 8.3 \times 1.2 + 10^5 \times 25 \times 10^{-3} \times 4 \times 10^{-4}$$

$$= 25.9$$

11. (c) $P_1 V_1^\gamma = P_2 V_2^\gamma$

$$\gamma = \frac{5}{3}$$

$$PV^{5/3} = P_2 \times (27V)^{5/3}$$

$$P_2 = \frac{P}{(27)^{5/3}} = \frac{P}{3^5}$$

$$\Delta U \Rightarrow nC_v\Delta T = \frac{nR\Delta T}{(\gamma - 1)}$$

$$= \frac{P_2 V_2 - P_1 V_1}{(\gamma - 1)}$$

$$\Rightarrow \frac{\frac{P}{3^5} \times 3^3 \cdot V - PV}{\frac{5}{3} - 1} \Rightarrow \frac{-PV \left\{1 - \frac{1}{9}\right\}}{\frac{2}{3}}$$

$$= -PV \times \frac{8}{9} \times \frac{3}{2}$$

$$\Rightarrow -\frac{4}{3}PV$$

12. (c) For a spring, $k\ell = \text{constant}$

$$k_1\ell = k_2 \frac{\ell}{2} \Rightarrow k_2 = 2k_1$$

$$f = \frac{1}{2\pi} \sqrt{\frac{k}{m}}$$

$$\Rightarrow \frac{f_1}{f_2} = \sqrt{\frac{k_1}{k_2}}$$

$$\Rightarrow \frac{f_1}{f_2} = \sqrt{\frac{1}{2}} \Rightarrow \frac{f_2}{f_1} = \sqrt{2}$$

$$\begin{aligned} 13. (d) \frac{hc}{\lambda} &= \frac{15 \times 10^3}{2.5 \times 10^{22}} \\ \therefore \lambda &= \frac{6.6 \times 10^{-34} \times 3 \times 10^8 \times 2.5 \times 10^{22}}{15 \times 10^3} \\ \lambda &= 3.3 \times 10^{-7} \end{aligned}$$

$$\begin{aligned} 14. (d) \vec{\tau} &= \vec{M} \times \vec{B} \\ \Rightarrow 100\sqrt{2} \times 3.14 \times 4 \times 10^{-4} \left[\left(\frac{\hat{k} + \hat{i}}{\sqrt{2}} \right) \times (3\hat{i} + 2\hat{k}) \right] B_0 \\ \Rightarrow 16 \times 3.14 \times [+3\hat{j} - 2\hat{j}] \times 10^{-5} \\ \Rightarrow 50.24 \times 10^{-5} \hat{j} \end{aligned}$$

$$\begin{aligned} 15. (b) \varepsilon &= -\frac{L di}{dt} = -\mu_0 n^2 A \ell \frac{di}{dt} \\ &= -4\pi \times 10^{-7} \times 10^6 \times 5 \times 10^{-4} \times 0.3 \times \frac{2}{\pi} \\ &\Rightarrow -12 \times 10^{-5} \end{aligned}$$

$$\begin{aligned} 16. (b) \vec{r} &= \vec{r}_2 - \vec{r}_1 \Rightarrow -\hat{i} - 2\hat{j} - 2\hat{k} \\ \text{Force} &\Rightarrow \frac{kq_1 q_2}{r^3} \vec{r} \\ &\Rightarrow \frac{9 \times 10^9 \times (3)(+4) \times 10^{-12}}{27} (\hat{i} + 2\hat{j} + 2\hat{k}) \\ &\Rightarrow 4 \times 10^{-3} [\hat{i} + 2\hat{j} + 2\hat{k}] \end{aligned}$$

$$\begin{aligned} 17. (a) 1(AO \times \hat{j}) &\Rightarrow (OB \times \hat{j}) \times 1.5 \\ \frac{2\hat{k}}{\sqrt{13}} &= \frac{3}{2} \left[\frac{C\hat{k}}{\sqrt{C^2 + 16}} \right] \\ \therefore \frac{4}{13} &= \frac{9C^2}{4(C^2 + 16)} \Rightarrow 16C^2 + 256 = 117C^2 \\ 256 &= 101C^2 \\ C &= 1.59 \approx 1.6 \end{aligned}$$

$$18. (d) 2K_1 = \frac{2hc}{\lambda_1} - 2\phi \quad \dots(i)$$

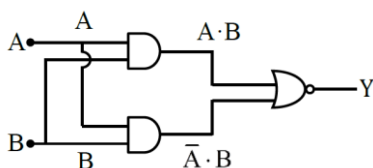
$$K_2 = \frac{2hc}{\lambda_1} - \phi \quad \dots(ii)$$

Solving (i) & (ii), we get

$$K_2 - 2K_1 = \phi$$

$$\begin{aligned} 19. (c) k_\alpha &= \left(\frac{A-4}{A} \right) Q \\ \therefore \frac{k_{\alpha_1}}{k_{\alpha_2}} &= \frac{196}{200} \times \frac{212}{208} = \frac{2597}{2600} \end{aligned}$$

20. (b)



$$A \cdot B + \bar{A} \cdot B \Rightarrow (A + \bar{A}) \cdot B = B$$

$$21. (2) RC = \rho K \epsilon_0$$

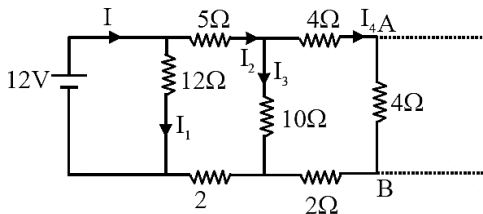
$$K = \frac{RC}{\rho \epsilon_0} = \frac{17.7 \times 10^{14} \times 10^{-6}}{10^{13} \times 8.85 \times 10^{-12}} = 2 \times 10^7$$

$$22.(203) \theta = \frac{1.22\lambda}{b}$$

$$3 \times 10^{-7} = \frac{1.22 \times 5 \times 10^{-7}}{b} \Rightarrow b = 203.33 \text{ cm}$$

$$23.(2) 5 \left(\frac{2\pi m}{qB} \right) v \Rightarrow \frac{5 \times 2 \times \pi \times 5 \times 10^{-6} \times 2 \times 10^{-2}}{5\pi \times 10^{-6} \times 0.1} = 2 \text{ m}$$

24.(200) At steady state,



$$I = \frac{E}{R_{eq}} = \frac{12}{6} = 2 \text{ A}$$

$$I_1 = I_2 = 1 \text{ A}$$

$$I_3 = I_4 = \frac{1}{2} \text{ A}$$

$$V_{AB} = \frac{1}{2} \times 4 = 2 \text{ V}$$

Charge on capacitor

$$Q = CV_{AB} = 100 \times 2 = 200 \mu\text{C}$$

$$25.(9) S = u + \frac{a}{2}(2n - 1) \Rightarrow 104 = 5 + \frac{a}{2}(9) \Rightarrow a = \frac{22 \text{ m}}{\text{s}^2}$$

$$129 = 12 + \frac{a}{2} \times 9 \Rightarrow a = 26 \text{ m/s}^2$$

$$\frac{p_1}{p_2} = \frac{(5 + 22 \times 10)3.4}{(12 + 26 \times 10)(2.5)} = \frac{225 \times (3.4)}{272 \times 2.5} = \frac{9}{8}$$

26. (c) (A) No. of atoms in 1.8 mg water

$$= \frac{1.8 \times 10^{-3}}{18} \times N_A \times 3 = 3 \times 10^{-4} N_A$$

(B) No. of atoms in 9.8 mg H_2SO_4

$$= \frac{9.8 \times 10^{-3}}{98} \times N_A \times 7 = 7 \times 10^{-4} \times N_A$$

(C) No. of atoms in 1.8 mg carbon

$$= \frac{1.8 \times 10^{-3}}{12} \times N_A = 1.5 \times 10^{-4} \times N_A$$

(D) No. of atoms in 5.85 mg NaCl

$$= \frac{5.85 \times 10^{-3}}{58.5} \times N_A \times 2 = 2 \times 10^{-4} N_A$$

27. (a) (I) Let wt of solution = 100 g

wt of methanol = 30 g

wt of CCl_4 = 100 - 30 = 70 g

$$\text{mole fraction of } \text{CCl}_4 = \frac{n_{\text{CCl}_4}}{n_{\text{CCl}_4} + n_{\text{CH}_3\text{OH}}} = \frac{\frac{70}{154}}{\frac{70}{154} + \frac{30}{32}} = 0.3267 \approx 0.33$$

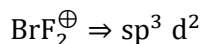
(II) CCl_4 & methanol is mixture of positive deviation from Raoult's law.

So, statement (I) & statement (II) are correct.

28. (a) $\text{BrF}_2^+ \Rightarrow \text{sp}^3$

[Bond pair = 2, lone pair = 2]

$\Rightarrow$ bent



[Bond pair = 4, lone pair = 2]

$\Rightarrow$ square planar

29.(c) (A) K_b for water = $0.512 \frac{^\circ\text{C-Kg}}{\text{mol}}$

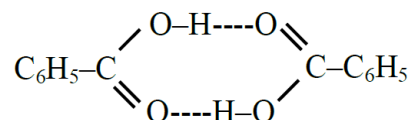
$$K_f \text{ for water} = 1.86 \frac{^\circ\text{C-Kg}}{\text{mol}}$$

So, $K_b < K_f$ for water

(B) Since, $K_b < K_f$ so $\Delta T_b < \Delta T_f$ [same equimolal solution]

(C) Osmotic pressure is used to determine molar mass of proteins & polymers

(D) dimerized form of benzoic acid in benzene



30. (c) $x_2 + y_2 \rightleftharpoons 2xy, K' = \frac{1}{K_1} = \frac{1}{2.5 \times 10^5} = \frac{1}{25 \times 10^4}$

$$\frac{1}{2}x_2 + \frac{1}{2}y_2 \rightleftharpoons xy, K'' = \left(\frac{1}{25 \times 10^4}\right)^{1/2} = \frac{1}{5 \times 10^2} \quad \dots(1)$$

$$xy + \frac{1}{2}z_2 \rightleftharpoons xyz, K_2 = 5 \times 10^{-3} \quad \dots(2)$$

On adding equation (1) & (2)

$$\frac{1}{2}x_2 + \frac{1}{2}y_2 + \frac{1}{2}z_2 \rightleftharpoons xyz,$$

$$K_3 = \frac{1}{5 \times 10^2} \times 5 \times 10^{-3} = 1 \times 10^{-5}$$

31. (b) (i) $\text{Fe}^{+2}(\text{aq}) + 2\text{e}^- \rightarrow \text{Fe}(\text{s}) \dots \dots \Delta G_1^\circ$

(ii) $\text{Fe}^{+3}(\text{aq}) + 3\text{e}^- \rightarrow \text{Fe}(\text{s}) \dots \dots \Delta G_2^\circ$

(iii) $\text{Fe}^{+3}(\text{aq}) + \text{e}^- \rightarrow \text{Fe}^{+2}(\text{aq}) \dots \dots \Delta G_3^\circ$

iii = ii - i

$$-1 \times F \times E^\circ_{\text{Fe}^{3+}/\text{Fe}^{2+}} = -3 \times F \times y - (-2 \times F \times x)$$

$$E^\circ_{\text{Fe}^{3+}/\text{Fe}^{2+}} = 3y - 2x$$

32.(c) $\text{S-I} : \ln\left(\frac{K_2}{K_1}\right) = \frac{12.6 \times 10^3}{\left(\frac{8.314}{4.2}\right)} \left[\frac{10}{298 \times 308}\right]$

$$\Rightarrow \ln\left(\frac{K_2}{K_1}\right) = 0.693$$

$$\Rightarrow \frac{K_2}{K_1} = 2$$

S-II : For 1st order

$$t_{1/2} = \frac{\ln 2}{k}; t_{1/2} \propto [A]^0 = \text{constant.}$$

Statement-I is true but statement-II is false.

33. (a) $2 \text{ s}^2 \Rightarrow \text{Be}$

(B) $2 \text{ s}^2 2 \text{ p}^1 \Rightarrow \text{B}$

(C) $2 \text{ s}^2 2 \text{ p}^3 \Rightarrow \text{N}$

(D) $2 \text{ s}^2 2 \text{ p}^6 \Rightarrow \text{Ne}$

IE order : $\text{B} < \text{Be} < \text{N} < \text{Ne}$

34. (c) (A) Reducing nature : $\text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3$

(B) Lone pair definition tendency :

$$\text{NH}_3 > \text{PH}_3 > \text{AsH}_3 > \text{SbH}_3 > \text{BiH}_3$$

(C) Stability of Hydrides :

$$\text{NH}_3 > \text{PH}_3 > \text{AsH}_3 > \text{SbH}_3 > \text{BiH}_3$$

(D) HEH bond angle : $\text{NH}_3 > \text{PH}_3 > \text{AsH}_3 > \text{SbH}_3 > \text{BiH}_3$

35. (a) Ti^{+4} is colourless due to d^0 configuration

$\therefore$ statement 1 is correct

$La^{+3}, Yb^{+2}, Lu^{+3}, Ce^{+4}, Al^{+3} \& Lr^{+3}$ all are diamagnetic

36. (b) First row transition metals utilize 3d and 4s electrons for bonding on the catalyst surface.

$\therefore$ Statement 1 is incorrect

Adsorption increases reactant concentration on the catalyst surface and weakens bonds facilitating formation of activated complex and lowering activation energy.

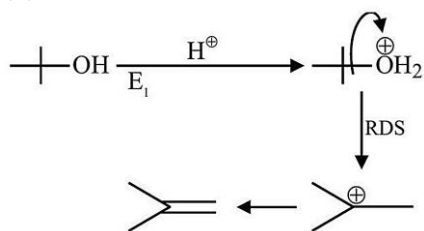
Statement 2 is incorrect

37. (c) Statement I : True

Statement II : False

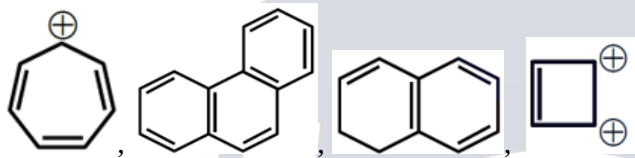
Reason - As vapors rise, they become richer in the lower-boiling component, not high-boiling component.

38. (b)

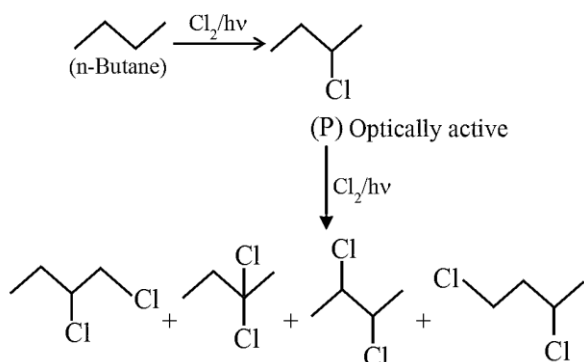


Product obtained without rearrangement.

39. (b) Aromatic compounds are



40. (b)



Total dichloro products (excluding stereoisomers) = 4

41. (a) Statement I : True

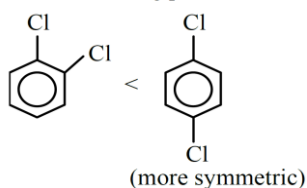
Order of boiling point :

$CH_3CH_2CH_2I > CH_3CH_2I > CH_3I$

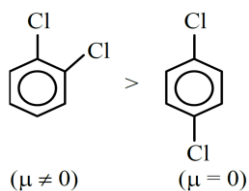
Vander Waals forces increases due to increase in molecular weight

Statement 2: True

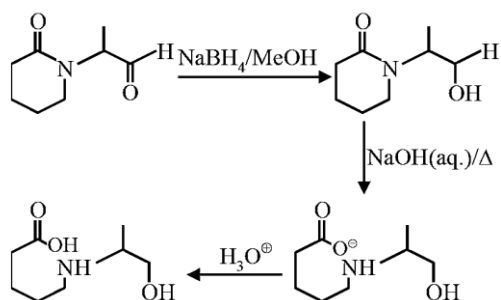
Order of melting point :-



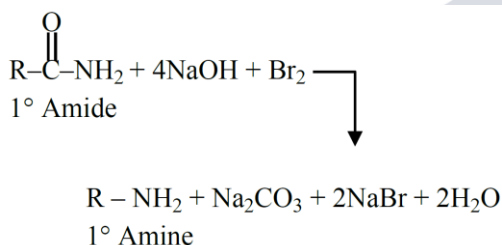
Boiling point order :-
(Boiling point $\propto \mu$)



42. (c)



43. (c) Hoffmann bromamide degradation reaction -



Statement (A) - False

Statement (B) - False

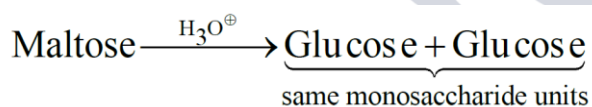
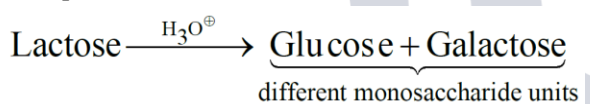
Statement (C) - True

Statement (D) - False

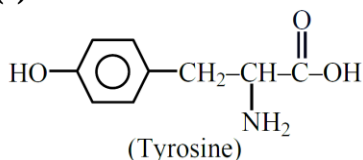
Statement (E) - True

44. (b) The monosaccharide units obtained from hydrolysis of oligosaccharides may be or may not be same.

Example :



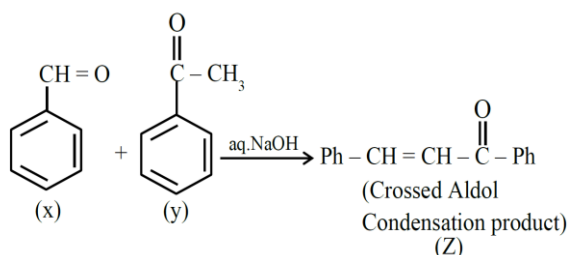
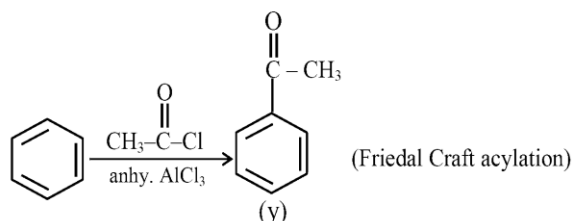
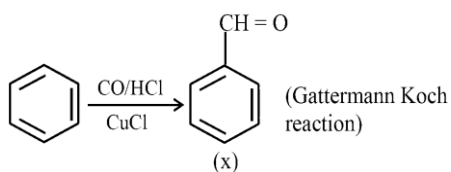
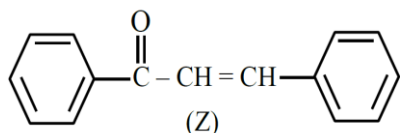
45. (c)



Tyrosine amino acid will give violet coloured complex with neutral FeCl_3 solution because it contains phenolic group.

46.(6) $[\text{MnBr}_4]^{2-}$, $[\text{NiCl}_4]^{2-}$, $[\text{CoF}_6]^{3-}$, $[\text{Mn}(\text{CN})_6]^{3-}$, $[\text{Ti}(\text{CN})_6]^{3-}$, $[\text{Cu}(\text{H}_2\text{O})_6]^{2+}$ are paramagnetic.

47. (16)


Total number of pi (π) electrons in 'z' = 16π electrons

48. (3) $E_{\text{photon}} = \frac{hc}{\lambda}$

$$\Rightarrow \frac{E_1}{E_2} = \frac{\lambda_2}{\lambda_1}$$

$$\Rightarrow \frac{E_1}{E_2} = \frac{6000}{2000}$$

$$\Rightarrow \frac{E_1}{E_2} = 3$$

49. (1126) $\Delta_r H^\circ = \Sigma \Delta H_f^\circ (\text{product}) - \Sigma \Delta H_f^\circ (\text{Reactant})$

$$\Delta_r H^\circ = 2 \times \Delta H_f^\circ (\text{SO}_2, \text{g}) + 2 \times \Delta H_f^\circ (\text{H}_2\text{O}, \ell) - 2 \times \Delta H_f^\circ (\text{H}_2\text{S}, \text{g})$$

$$\Delta_r H^\circ = 2 \times (-297) + 2 \times (-286) - 2(-20.1)$$

$$= -594 - 572 + 40.2$$

$$\Delta_r H^\circ = -1125.8 \text{ kJ/mol.}$$

$$|\Delta_r H^\circ| = 1125.8 \text{ kJ/mol.}$$

50. (4) $\text{CaCO}_3(\text{s}) \rightleftharpoons \text{CaO}(\text{s}) + \text{CO}_2(\text{g}), K_{P_1} = 0.08 \text{ atm}$

$$K_{P_1} = P_{\text{CO}_2} = 0.08 \text{ atm}$$

$$\text{C}(\text{s}) + \text{CO}_2(\text{g}) \rightleftharpoons 2\text{CO}(\text{g}), K_{P_2} = 2$$

$$K_{P_2} = \frac{P_{\text{CO}}^2}{P_{\text{CO}_2}} = 2$$

$$\frac{P_{\text{CO}}^2}{0.08} = 2$$

$$P_{\text{CO}}^2 = 16 \times 10^{-2}$$

$$P_{\text{CO}} = 4 \times 10^{-1}$$

51. (a) Reflexive If $(a, a) \in R$

$$\Rightarrow 1 + a^2 > 0$$

 $\Rightarrow$ It is reflexive relation

Symmetric If $1 + ab > 0 \Rightarrow 1 + ba > 0$

i.e. If $(a, b) \in R$ and $(b, a) \in R$

Therefore it is symmetric relation

Transitive $\because (-2, 0) \in R \& (0, 2) \in R$

But $(-2, 2) \notin R$

Hence it is not transitive

$\Rightarrow$ It is not an equivalence relation.

Check Number of elements in R

$\because 1 + ab > 0$

a & b each can be filled by 5 ways

$\therefore$ total elements = 25 But out of these

$(-2, -1), (-2, 2), (-1, 2), (1, -2), (2, -2), (2, -1)$

$(-1, 1) \& (1, -1)$ are not in relation.

52. (b) $|z - (3 + 5i)| + |z - (5 + 11i)| = 4\sqrt{5}$

So, $S_1(3, 5)$ & $S_2(5, 11)$ are two foci of ellipse, where $2ae = S_1 S_2 = 2\sqrt{10}$ & $2a = 4\sqrt{5}$

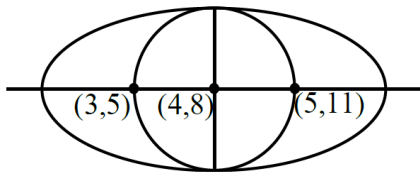
$\Rightarrow a = 2\sqrt{5}, e = \frac{1}{\sqrt{2}} \& b^2 = a^2(1 - e^2)$ gives $b^2 = 10$

& center of an ellipse is mid point of both focus

$\therefore$ center $(4, 8)$

Equation of circle is $|Z - (4 + 8i)| = \sqrt{10}$

$\Rightarrow$ Ellipse & given circle are concentric & radius of given circle is equal to length of minor axis of given ellipse



$\Rightarrow$ Two points which are common to both.

53. (c) $D = \begin{vmatrix} 1 & 1 & 1 \\ 1 & 2 & 5 \\ 2 & 3 & \lambda \end{vmatrix} = \lambda - 6$

$D_1 = \begin{vmatrix} 6 & 1 & 1 \\ 10 & 2 & 5 \\ \mu & 3 & \lambda \end{vmatrix} = 2\lambda + 3\mu - 60$

$D_2 = \begin{vmatrix} 1 & 6 & 1 \\ 1 & 10 & 5 \\ 2 & \mu & \lambda \end{vmatrix} = 4(\lambda - \mu + 10)$

$D_3 = \begin{vmatrix} 1 & 1 & 6 \\ 1 & 2 & 10 \\ 2 & 3 & \mu \end{vmatrix} = \mu - 16$

For infinite many solution

$D = D_1 = D_2 = D_3 = 0$

$\Rightarrow \lambda = 6 \& \mu = 16$

$\Rightarrow \lambda + \mu = 22$

54. (c) $A = \begin{bmatrix} \alpha & 1 & 2 \\ 2 & 3 & 0 \\ 0 & 4 & 5 \end{bmatrix}$ & $|A| = 5\alpha + 6$

Now $\text{adj}A = \begin{bmatrix} 15 & 3 & -6 \\ -10 & 5\alpha & 4 \\ 8 & -4\alpha & 3\alpha - 2 \end{bmatrix}$

$\therefore B = \begin{bmatrix} 1 & 0 & 0 \\ 0 & -5\alpha & 0 \\ 0 & 4\alpha & -2\alpha \end{bmatrix} + \text{adj}(A)$

$$B = \begin{bmatrix} 16 & 3 & -6 \\ -10 & 0 & 4 \\ 8 & 0 & \alpha - 2 \end{bmatrix}$$

$$\&|B| = 6(5\alpha + 6) = 66 \text{ (given in question)}$$

$$\Rightarrow \alpha = 1$$

$$\Rightarrow |A| = 21$$

$$\therefore \det(\text{adj}A) = |A|^{(n-1)^2}$$

$$\text{Here } n = 3$$

$$= 21^2$$

$$= 441$$

$$55. (a) a = (3 + 8 + 13 \dots \text{upto } 20 \text{ terms}) + (4 + 9 + 14 + \dots \text{upto } 20 \text{ terms})$$

$$= \frac{20}{2} [6 + 19 \times 5] + \frac{20}{2} (8 + 19 \times 5)$$

$$= 2040$$

$$(\tan \beta)^{\frac{2040}{1020}} = \tan^2 \beta$$

$$x^2 + x - 2 = 0 < \frac{1}{2}$$

$$\Rightarrow \tan^2 \beta = 1; \sin^2 \beta = \frac{1}{2} \& \cos^2 \beta = \frac{1}{2}$$

$$\sin^2 \beta + 3\cos^2 \beta = \frac{1}{2} + \frac{3}{2} = \frac{4}{2} = 2$$

$$56. (b) P(\text{Bus}) = P(B) = \frac{2}{5} P\left(\frac{\text{late}}{B}\right) = \frac{1}{5}$$

$$P(\text{Scooter}) = P(S) = \frac{1}{5} P\left(\frac{\text{late}}{S}\right) = \frac{1}{3}$$

$$P(\text{Car}) = P(C) = \frac{2}{5} P\left(\frac{\text{late}}{C}\right) = \frac{1}{4}$$

$$P\left(\frac{B}{\text{late}}\right) = \frac{\frac{2}{5} \times \frac{1}{5}}{\frac{2}{5} \times \frac{1}{5} \times \frac{1}{5} \times \frac{1}{3} \times \frac{2}{5} \times \frac{1}{4}} = \frac{12}{37}$$

$$57. (c) \text{ Given } \bar{x} = 1, \sigma_1^2 = 13$$

$$\bar{y} = 2, \sigma_2^2 = 1$$

$$\text{Combined variance} = \frac{n_1 \sigma_1^2 + n_2 \sigma_2^2}{n_1 + n_2} + \frac{n_1 n_2}{(n_1 + n_2)^2} (\bar{x} - \bar{y})^2$$

$$= \frac{4 \times 13 + 6 \times 1}{10} + \frac{4 \times 6}{10^2} (2 - 1)^2$$

$$\frac{58}{10} + \frac{24}{100} = \frac{580 + 24}{100} = \frac{604}{100} = 6.04$$

$$58. (c) (1+x)^{12} = {}^{12}C_0 + {}^{12}C_1 x + {}^{12}C_2 x^2 + \dots + {}^{12}C_{12} x^{12} \text{ integrating both sides}$$

$$\left(\frac{(1+x)^{13}}{13}\right)_{-2}^2 = \left({}^{12}C_0 x + {}^{12}C_1 \frac{x^2}{2} + {}^{12}C_2 \frac{x^3}{3} + \dots + {}^{12}C_{12} \frac{x^{13}}{13}\right)_{-2}^2$$

$$\frac{3^{13}}{13} + \frac{1}{13} = \left({}^{12}C_0 2 + {}^{12}C_1 \frac{2^2}{2} + {}^{12}C_2 \frac{2^3}{3} + \dots + {}^{12}C_{12} \frac{2^{13}}{13}\right)$$

$$-\left(-{}^{12}C_0 \cdot 2 + {}^{12}C_1 \frac{2^2}{2} - {}^{12}C_2 \cdot \frac{2^3}{3} + \dots + {}^{12}C_{12} \cdot \frac{2^{13}}{13}\right)$$

$$= 2 \left(2 \cdot {}^{12}C_0 + {}^{12}C_2 \cdot \frac{2^3}{3} + {}^{12}C_4 \cdot \frac{2^5}{5} + \dots + {}^{12}C_{12} \cdot \frac{2^{13}}{13}\right)$$

$$3^{13} + 1 = 26 \left(2 + {}^{12}C_{12} \cdot \frac{2^3}{3} + {}^{12}C_4 \cdot \frac{2^5}{5} + \dots + {}^{12}C_{12} \cdot \frac{2^{13}}{13}\right)$$

$$3^{13} - 51 = 26 \left({}^{12}C_2 \cdot \frac{2^3}{3} + {}^{12}C_4 \cdot \frac{2^5}{5} + \dots + {}^{12}C_{12} \cdot \frac{2^{13}}{13}\right)$$

$$\therefore \alpha = 51$$

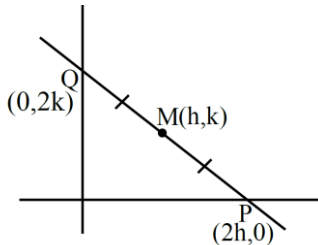
59. (b)

$$\begin{array}{ccc} B_1 & B_1 & B_1 \\ 1 & 1 & 2 \end{array}$$

$$\text{No. of ways} = \frac{4!}{2!2!} \times 3! = 6 \times 6 = 36$$

60. (b) Point of intersection of the lines

$$3x + 4y = 1 \text{ and } 4x + 3y = 1 \text{ is } \left(\frac{1}{7}, \frac{1}{7}\right)$$



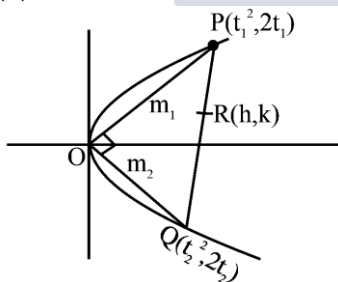
$$\text{Let the line be } \frac{x}{2h} + \frac{y}{2k} = 1$$

$$\text{Satisfy } \left(\frac{1}{7}, \frac{1}{7}\right)$$

$$\frac{1}{14h} + \frac{1}{14k} = 1$$

$$\frac{1}{x} + \frac{1}{y} = 14$$

61. (b)



$$t_1 t_2 = -4$$

$$2h = t_1^2 + t_2^2$$

$$2k = 2(t_1 + t_2)$$

$$2h = k^2 - 2t_1 t_2$$

$$2h = k^2 + 8$$

$$k^2 = 2h - 8$$

$$= 2(h - 4)$$

$$\text{L.R.} = 2$$

62. (a) $\frac{1}{2} < \frac{6}{11} < \frac{1}{\sqrt{2}}$

$$\sin^{-1}\left(\frac{1}{2}\right) < \sin^{-1}\left(\frac{6}{11}\right) < \sin^{-1}\left(\frac{1}{\sqrt{2}}\right)$$

$$\frac{\pi}{6} < 3\sin^{-1}\left(\frac{6}{11}\right) < \frac{\pi}{4}$$

$$\frac{\pi}{2} < \alpha < \frac{3\pi}{4} \therefore \cos \alpha < 0$$

$$0 < \frac{4}{9} < \frac{1}{2}$$

$$\frac{\pi}{3} < \cos^{-1}\left(\frac{4}{9}\right) < \frac{\pi}{2}$$

$$\pi < 3\cos^{-1}\left(\frac{4}{9}\right) < \frac{3\pi}{2}$$

$$\pi < \beta < \frac{3\pi}{2}$$

$$\text{Now } \frac{3\pi}{2} < \alpha + \beta < \frac{9\pi}{4}$$

$$\therefore \cos(\alpha + \beta) > 0$$

63.(d) $f(x) = e^{|\sin x|} - |x|$ is non-differentiable at $x = \pi$ and for $x \in \left(-\pi, \frac{-\pi}{2}\right)$

$$f(x) = e^{-\sin x} + x$$

$$f'(x) = -e^{-\sin x} \cos x + 1$$

$$\text{for } x \in \left(-\pi, \frac{-\pi}{2}\right) \cos x < 0$$

$$\therefore f'(x) > 0$$

$$\therefore f(x) \text{ is increasing}$$

64. (b) $2(\vec{a} \times \vec{b}) - 3(\vec{c} \times \vec{b}) = \vec{0}$

$$(2\vec{a} - 3\vec{c}) \times \vec{b} = \vec{0}$$

$$\vec{b} \parallel (2\vec{a} - 3\vec{c})$$

$$2\vec{a} - 3\vec{c} = \lambda \vec{b}$$

$$\vec{c} = \frac{2\vec{a} - \lambda \vec{b}}{3} \quad \dots(1)$$

$$\vec{a} \cdot \vec{c} = 15$$

$$\left(\frac{2\vec{a} - \lambda \vec{b}}{3}\right) \cdot \vec{a} = 15$$

$$\frac{2|\vec{a}|^2 - \lambda \vec{b} \cdot \vec{a}}{3} = 15$$

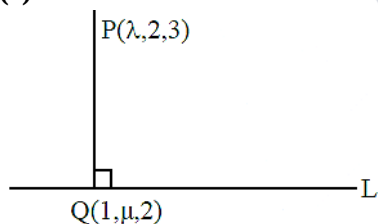
$$2(26) - \lambda(40 - 2 - 3) = 45$$

$$\lambda = \frac{1}{5}$$

$$\Rightarrow \vec{c} \cdot (\hat{i} + \hat{j} - 3\hat{k}) = \frac{(2(4\hat{i} - \hat{j} + 3\hat{k}) - \frac{1}{5}(10\hat{i} + 2\hat{j} - \hat{k})) \cdot (\hat{i} + \hat{j} - 3\hat{k})}{3}$$

$$= \frac{2(4 - 1 - 9) - \frac{1}{5}(10 + 2 + 3)}{3} = -5$$

65. (c)



$$\frac{x-4}{1} = \frac{y-9}{2} = \frac{z-5}{1}$$

$Q(1, \mu, 2)$ satisfies the line L

$$\therefore \frac{1-4}{1} = \frac{\mu-9}{2} = \frac{2-5}{1}$$

$$\Rightarrow -3 = \frac{\mu-9}{2} = -3 \Rightarrow \mu = 3$$

$$\text{D.R. of PQ: } \lambda - 1, 2 - \mu, 1$$

$$\text{Since } PQ \perp L \Rightarrow (\lambda - 1) \cdot 1 + 2(2 - \mu) + 1 \cdot 1 = 0$$

$$\Rightarrow \lambda - 2\mu + 4 = 0$$

$$\therefore \lambda = 2$$

Now lines : $\frac{x-1}{2} = \frac{y-2}{3} = \frac{z+4}{6}$ &
 $\frac{x-2}{2} = \frac{y-3}{3} = \frac{z+5}{6}$

i.e. $\vec{r} = (\hat{i} + 2\hat{j} - 4\hat{k}) + t_1(2\hat{i} + 3\hat{j} + 6\hat{k})$

& $\vec{r} = (2\hat{i} + 3\hat{j} - 5\hat{k}) + t_2(2\hat{i} + 3\hat{j} + 6\hat{k})$

Distance

$$= \frac{|((\hat{i} + 2\hat{j} - 4\hat{k}) - (2\hat{i} + 3\hat{j} - 5\hat{k})) \times (2\hat{i} + 3\hat{j} + 6\hat{k})|}{|2\hat{i} + 3\hat{j} + 6\hat{k}|}$$

$$= \frac{|-9\hat{i} + 8\hat{j} - \hat{k}|}{7} = \frac{\sqrt{146}}{7}$$

66. (c) $I = \int_0^2 \sqrt{\frac{x(x^2+x+1)}{(x+1)(x^2-x+1)(x^2+x+1)}} dx$

$$= \int_0^2 \sqrt{\frac{x}{x^3+1}} dx$$

$$(x^{3/2} = t \Rightarrow \frac{3}{2}x^{1/2}dx = dt)$$

$$= \frac{2}{3} \int_0^{2^{3/2}} \frac{dt}{\sqrt{t^2+1}}$$

$$= \frac{2}{3} \left(\ln(t + \sqrt{t^2+1}) \right)_0^{2^{3/2}}$$

$$= \frac{2}{3} \ln(2^{3/2} + 3)$$

67. (a) $x\sqrt{1-x^2}dy + (y\sqrt{1-x^2} - x\cos^{-1}x)dx = 0$

$$\Rightarrow \frac{dy}{dx} + \frac{y}{x} = \frac{\cos^{-1}x}{\sqrt{1-x^2}}$$

it is L.D.E.

$$\therefore \text{I.F.} = e^{\int \frac{1}{x} dx} = e^{\ln x} = x$$

$$\therefore \text{solution is } y \cdot x = \int \frac{x\cos^{-1}x}{\sqrt{1-x^2}} dx + C$$

$$y \cdot x = I_1 + C \quad \dots(1)$$

$$I_1 = \int \frac{x\cos^{-1}x}{\sqrt{1-x^2}} dx$$

$$\text{Let } \cos^{-1}x = t \Rightarrow x = \cos t$$

$$-\frac{1}{\sqrt{1-x^2}} dx = dt$$

$$I_1 = \int -t \cdot \cot t dt$$

$$= -(t \sin t + \cos t)$$

$$I_1 = -(\sqrt{1-x^2} \cdot \cos^{-1}x + x) \quad \dots(2)$$

$$\therefore y \cdot x = -(\sqrt{1-x^2} \cdot \cos^{-1}x + x) + C$$

$$\therefore \lim_{x \rightarrow 1^-} y(x) = 1$$

$$\therefore 1 = -(0 + 1) + C$$

$$C = 2$$

$$\therefore y \cdot x = -\sqrt{1-x^2} \cdot \cos^{-1}x - x + 2$$

$$\left(\text{put } x = \frac{1}{2} \right)$$

$$y\left(\frac{1}{2}\right) \times \frac{1}{2} = -\frac{\sqrt{3}}{2} \times \frac{\pi}{3} - \frac{1}{2} + 2$$

$$\Rightarrow y\left(\frac{1}{2}\right) = 3 - \frac{\pi}{\sqrt{3}}$$

$$68. (a) f^2(x) = f(f(x)) = \frac{\frac{x-1}{x+1}-1}{\frac{x-1}{x+1}+1} = \frac{-1}{x}$$

$$f^3(x) = f(f^2(x)) = f\left(\frac{-1}{x}\right) = \frac{1+x}{1-x}$$

$$f^4(x) = f(f^3(x)) = f\left(f\left(\frac{-1}{x}\right)\right) = x$$

$$f^5(x) = f(f^4(x)) = f(x)$$

$$f^6(x) = f(f^5(x)) = f(f(x)) = \frac{-1}{x}$$

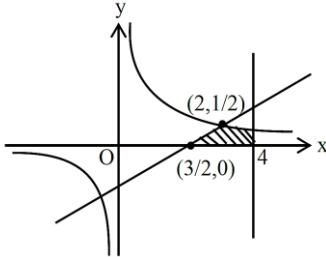
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$$f^{26}(x) = \frac{-1}{x}$$

$$g(x) = \frac{1}{x}$$



$$x - 3/2 = 1/x \Rightarrow 2x^2 - 3x - 2 = 0$$

$$\Rightarrow x = -1/2, 2$$

$$\text{Req. area} = \int_2^4 \frac{1}{x} dx + \frac{1}{2} \times \frac{1}{2} \times \frac{1}{2}$$

$$= \frac{1}{8} + (\ln x)_2^4 = \frac{1}{8} + \ln 2$$

$$69. (d) \lim_{x \rightarrow \frac{\pi}{2}} \frac{b(1 - \cos(\frac{\pi}{2} - x))}{4(\frac{\pi}{2} - x)^2} = \frac{1}{3} \Rightarrow b = \frac{8}{3}$$

$$\int_0^2 |x^2 + 2x - 3| dx = \int_0^2 |(x+3)(x-1)| dx$$

put $x - 1 = t$

$$\int_{-1}^1 |(t+4)t| dt = \int_{-1}^0 (-t^2 - 4t) dt + \int_0^1 (t^2 + 4t) dt$$

$$\left(\frac{-t^3}{3} - 2t^2\right)_{-1}^0 + \left(\frac{t^3}{3} + 2t^2\right)_0^1 = 4$$

70. (b) Given ellipse is vertical

$$\therefore f(3a + 15) > f(a^2 + 7a + 3)$$

$$\therefore f(x) \text{ is decreasing } \forall x \in \mathbb{R}$$

$$\Rightarrow 3a + 15 < a^2 + 7a + 3$$

$$\Rightarrow a^2 + 4a - 12 > 0$$

$$\Rightarrow (a + 2)^2 > 16$$

$$\Rightarrow |(a + 2)| > 4$$

$$\Rightarrow a > 2 \text{ or } a < -6$$

$$\Rightarrow a \in \mathbb{R} - [-6, 2]$$

$$\therefore \alpha = -6, \beta = 2$$

$$\therefore \alpha^2 + \beta^2 = 36 + 4 = 40$$

$$71. (2) \log_{(x+1)}(2x+3)(x+1) + \log_{(2x+3)}(x+1)^2 = 4$$

$$1 + \log_{(x+1)}(2x+3) + 2\log_{(2x+3)}(x+1) = 4$$

$$\log_{x+1}(2x+3) = t \quad t + \frac{2}{t} = 3$$

$$t^2 - 3t + 2 = 0 \Rightarrow t = 1, 2$$

$$\text{for } t = 1 \Rightarrow \log_{(x+1)}(2x+3) = 1 \Rightarrow 2x+3 = x+1$$

$$\Rightarrow x = -2 \text{ (rejected)}$$

$$\text{for } t = 2 \Rightarrow 2x+3 = x^2 + 2x + 1 \Rightarrow x^2 = 2$$

$$\Rightarrow x = \pm\sqrt{2}$$

$$\text{rejecting } x = -\sqrt{2}, \text{ we get } x = \sqrt{2}$$

$$\therefore \text{sum of squares of all the roots} = 2$$

72. (12) $\int_{\pi/6}^{\pi/4} \left(\cot\left(x - \frac{\pi}{3}\right) \cot\left(x + \frac{\pi}{3}\right) + 1 \right) dx$

$$\text{using } \cot(A-B) = \frac{\cot B \cot A + 1}{\cot B - \cot A}$$

$$\Rightarrow \int_{\pi/6}^{\pi/4} -\frac{1}{\sqrt{3}} \left[\cot\left(x - \frac{\pi}{3}\right) - \cot\left(x + \frac{\pi}{3}\right) \right] dx$$

$$\Rightarrow \frac{-1}{\sqrt{3}} \left[\ln \left| \frac{\sin\left(x - \frac{\pi}{3}\right)}{\sin\left(x + \frac{\pi}{3}\right)} \right| \right]_{\pi/6}^{\pi/4}$$

$$= \frac{-1}{\sqrt{3}} \left[\ln \left| \frac{\sin 15^\circ}{\sin 105^\circ} \right| - \ln \left| \frac{\sin 30^\circ}{\sin 90^\circ} \right| \right]$$

$$= \frac{-1}{\sqrt{3}} \left[\ln(\tan 15^\circ) - \ln\left(\frac{1}{2}\right) \right]$$

$$= \frac{-1}{\sqrt{3}} [\ln(2 - \sqrt{3}) + \ln 2]$$

$$\Rightarrow \frac{-1}{\sqrt{3}} \ln(4 - 2\sqrt{3}) = \frac{-2}{\sqrt{3}} \ln(\sqrt{3} - 1)$$

$$\therefore \alpha = \frac{-2}{\sqrt{3}} \therefore 9\alpha^2 = 12$$

73. (4) $L_2: \frac{x-3}{1} = \frac{y+1}{2} = \frac{z-4}{2} = t$

$$L_5: \frac{x-3}{2} = \frac{y-3}{2} = \frac{z-2}{1} = s$$

vector $\perp^{\text{ar}}$ to L_2 & L_3

$$\Rightarrow \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & 2 & 2 \\ 2 & 2 & 1 \end{vmatrix} = \hat{i}(-2) - \hat{j}(-3) + \hat{k}(-2)$$

$$\Rightarrow -2\hat{i} + 3\hat{j} - 2\hat{k}$$

$$\Rightarrow -(2\hat{i} - 3\hat{j} + 2\hat{k})$$

$$\text{Now, } L_1: \frac{x}{2} = \frac{y}{-3} = \frac{z}{2} = \ell$$

intersection of L_1 & L_2

$$\Rightarrow t+3 = 2\ell, 2t-1 = -3\ell, 2t+4 = 2\ell$$

$$t+3 = 2t+4$$

$$t = -1$$

$$\Rightarrow P \equiv (2, -3, 2)$$

$$\text{A point on } L_3 \text{ is } Q(2s+3, 2s+3, s+2)$$

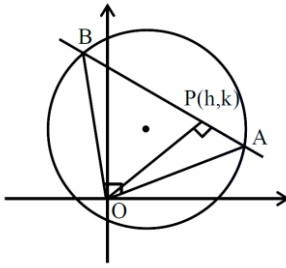
$$\Rightarrow PQ^2 = (2s+1)^2 + (2s+6)^2 + s^2 = 17$$

$$\Rightarrow 9s^2 + 28s + 20 = 0$$

$$\Rightarrow (9s+10)(s+2) = 0 \Rightarrow s = -2$$

$$\therefore Q \equiv (-1, -1, 0) \equiv (a, b, c) \Rightarrow (a+b+c)^2 = 4$$

74. (18)



$$C: x^2 + y^2 - 6x - 8y - 11 = 0$$

 E^{qn} of chord AB

$$y - k = -\frac{h}{k}(x - h)$$

$$hx + ky = h^2 + k^2 \quad \dots(2)$$

Homogenizing the eqⁿ of the circle

$$x^2 + y^2 - 6x \left(\frac{hx + ky}{h^2 + k^2} \right) - 8y \left(\frac{hx + ky}{h^2 + k^2} \right) - 11 \left(\frac{hx + ky}{h^2 + k^2} \right)^2 = 0$$

$$\text{coef. of } x^2 + \text{coef. of } y^2 = 0$$

$$1 - \frac{6h}{h^2 + k^2} - 11 \cdot \frac{h^2}{(h^2 + k^2)^2} + 1 - \frac{8k}{h^2 + k^2} - 11 \cdot \frac{k^2}{(h^2 + k^2)^2} = 0$$

$$2(h^2 + k^2) - 6h - 8k - 11 = 0$$

$$x^2 + y^2 - 3x - 4y - \frac{11}{2} = 0$$

$$\therefore \alpha + \beta + 2\gamma = 3 + 4 + 11 = 18$$

$$75. (395) \log_2 f(x) = \log_2 \left(\frac{2}{1 - \frac{1}{3}} \right) \cdot \log_3 \left(1 + \frac{f(x)}{f\left(\frac{1}{x}\right)} \right)$$

$$\log_2 f(x) = \log_2 3 \cdot \log_3 \left(1 + \frac{f(x)}{f\left(\frac{1}{x}\right)} \right)$$

$$\Rightarrow f(x) = 1 + \frac{f(x)}{f\left(\frac{1}{x}\right)}$$

$$\Rightarrow f(x)f\left(\frac{1}{x}\right) = f(x) + f\left(\frac{1}{x}\right)$$

$$\Rightarrow f(x) = 1 \pm x^n \Rightarrow f(6) = 37$$

$$\Rightarrow 1 \pm 6^n = 37 \Rightarrow 6^n = 36$$

$$\Rightarrow n = 2$$

$$\therefore f(x) = 1 + x^2$$

$$\sum_{n=1}^{10} (1 + n^2) = 10 + \frac{10 \cdot 11 \cdot 21}{6} = 395$$