

1. (a) $R_1 = 32 \times 10 = 320\Omega$

$$R_3 = \frac{R_y}{R_2} \times R_1 = \frac{40 \times 320}{80} = 160\Omega$$

$\therefore$ Colour code of R_3 be Brown, Blue, Brown.

2. (d) $Q = (B.E.)_R - (B.E.)_P$
 $= 20 \times 8.03 - (8 \times 7.07 + 12 \times 7.86)$
 $= 160.6 - (56.56 + 94.32)$

$\therefore Q = +9.72\text{meV}$

9.72 MeV released.

3. (a) $\therefore T = 2\pi \sqrt{\frac{1}{\mu B}}$

$$\frac{T_h}{T_c} = \sqrt{\frac{I_R \times \frac{\mu_c}{\mu_h}}{I_c \times \frac{\mu_c}{\mu_h}}}$$

$$= \sqrt{2 \times \frac{1}{2}} = 1$$

$T_h = T_c$

4. (c)

Heat loss = Heat gain

$$192 \times S(100 - 21.5) = (128 \times 0.394 + 240 \times 4.2)(21.5 - 8.4)$$

$$192 \times 78.5 \times S = 1058.432 \times 13.1$$

$$S = 0.91995 \text{ J/gK}^{-1}$$

$$S = 919.95 \text{ J/kgK}^{-1}$$

5. (a) $I = 2 \left[\frac{2}{5} MR^2 + M4R^2 \right] + M \frac{4R^2}{12}$

$$= MR^2 \left[\frac{1}{3} + \frac{4}{5} + 8 \right] = \frac{137}{15} MR^2$$

6. (b) $\frac{L\Delta T}{\Delta t} = 25$

$$\Rightarrow L = \frac{25 \times 1}{15} = \frac{5}{3}$$

$$\Delta U = \frac{1}{2} L (I_f^2 - I_i^2) = \frac{1}{2} \times \frac{5}{3} \times (25^2 - 10^2)$$

$$= \frac{5}{6} \times 525 = 4.75 \text{ J}$$

7. (b) $\frac{30R}{R+30} = 30 \times 0.95$

$$\Rightarrow R = 570\Omega$$

8. (d) $\tau = F \times 0.06 = 1.8 \times 0.012 \times 18 \times 10^{-6}$

$$F = 6.48 \times 10^{-5}$$

$$9. (a) |\vec{A} + \vec{B}| = n|\vec{A} - \vec{B}|$$

$$\Rightarrow A^2 + B^2 + 2AB\cos\theta$$

$$= n^2(A^2 + B^2 - 2AB\cos\theta)$$

$$\Rightarrow \cos\theta(1 + n^2) = \frac{2a^2(n^2 - 1)}{2a^2} [A = B = a]$$

$$\cos\theta = \frac{n^2 - 1}{n^2 + 1}$$

$$10. (c) n = \frac{16 \times 10^{-3} \times 10^{-4}}{10 \times 10 \times 10^{-19}} \approx 1.6 \times 10^{11}$$

$$(K.E.)_{\max} = (10 - 5)eV = 5eV$$

$$11. (d) K.E. - 3 = \vec{F} \cdot \vec{d}$$

$$K.E. = 3 + (3\hat{i} - 12\hat{j}) \times (4\hat{i})$$

$$K.E. = 3 + 12 = 15 \text{ J}$$

$$12. (a) \sqrt{(2d)^2 + (d)^2} - 2d = \frac{\lambda}{2}$$

$$\Rightarrow (\sqrt{5} - 2)d = \frac{\lambda}{2}$$

$$\Rightarrow d = \frac{\lambda}{2(\sqrt{5} - 2)}$$

$$13. (d) \frac{\mu_2}{v} - \frac{\mu_1}{u} = \frac{\mu_2 - \mu_1}{R}$$

$$\Rightarrow \frac{1.34}{V} - \frac{1}{-\infty} = \frac{0.34}{7.8}$$

$$\Rightarrow v = \frac{1.34 \times 7.8}{0.34} \text{ nm} = 3.074 \text{ cm}$$

$$14. (a)$$

$$R = \frac{P}{I^2} = \frac{4.4}{4 \times 10^{-6}} \Omega$$

$$P' = \frac{V^2}{R} = \frac{11 \times 11 \times 4 \times 10^{-6}}{4.4} \text{ W}$$

$$= 11 \times 10^{-5} \text{ W}$$

$$15. (c)$$

$$v = \pi R^2 h = \frac{\pi}{4} D^2 h$$

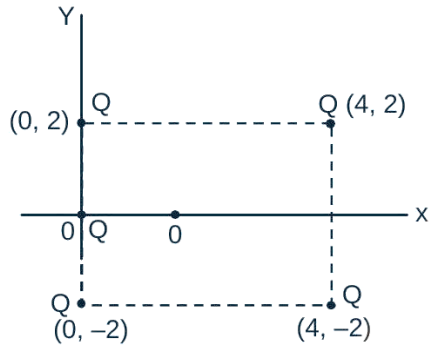
$$= 4260 \text{ cm}^2$$

$$\therefore \frac{\Delta v}{v} = 2 \frac{\Delta D}{D} + \frac{\Delta h}{h}$$

$$\begin{aligned}
 &= \left(2 \times \frac{0.1}{12.6} + \frac{0.1}{34.2} \right) v \\
 &= \frac{2 \times 426}{12.6} + \frac{426}{34.2} \\
 &= 67.61 + 12.459 = 80.075
 \end{aligned}$$

$$\therefore v = 4260 \pm 80 \text{ cm}^3$$

16. (b)



$$\begin{aligned}
 W &= VQ = \frac{1}{4\pi\epsilon_0} Q^2 \left[\frac{1}{2} + \frac{1}{2} + \frac{2}{2\sqrt{5}} \right] \\
 &\therefore \frac{Q^2}{4\pi\epsilon_0} \left[1 + \frac{1}{\sqrt{5}} \right]
 \end{aligned}$$

17. (d)

18. (a) $\frac{250\text{kHz}}{1.5\text{kHz}} = 13.33$

$\therefore$ Possible hormones

1,3,5,7,9,11,13

i.e 6 .

19. (a) $i_{10k} = \frac{50}{10k} = 5 \text{ mA}$

$$i_{5k} = \frac{120 - 50}{5k} = 14 \text{ mA}$$

$$i_2 = (14 - 5)\text{mA} = 9 \text{ mA}$$

20. (b) $\vec{E} = 10\hat{j}\cos(6x + 8z - 10ct)$

$$B_0 = \frac{E_0}{c} = \frac{10}{c}$$

$$W = 10C$$

$$\therefore \hat{E} \times \hat{B} = \hat{C}$$

$$\begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 0 & 1 & 0 \\ B_x & B_y & B_z \end{vmatrix} = \frac{6\hat{i} + 8\hat{j}}{10}$$

$$\Rightarrow B_z \hat{i} - 0\hat{j} - B_x \hat{k} = \frac{3}{5}\hat{i} + \frac{4}{5}\hat{j}$$

$$B_x = \frac{3}{5}, B_y = 0, B_z = \frac{4}{5}$$

$$\therefore \vec{B} = \frac{1}{C}(-8\hat{i} + 6\hat{k})\cos(6x + 8z + 10ct)$$

21. (d)

$$\therefore F \propto \frac{1}{r^3}$$

Required force = 27 F.

22. (d)

23. (b)

$$W = nR\Delta T$$

$$= \frac{1}{2} \times 8.31 \times 70$$

$$= 290.85 \text{ J}$$

24. (b)

$$W = \frac{Q^2}{2c} - \frac{Q^2}{2ck}$$

$$= \frac{Q^2}{2c} \left[1 - \frac{1}{k} \right]$$

$$= \frac{1}{2} \times 12 \times 100 \text{ pJ} \left(1 - \frac{1}{6.5} \right)$$

$$= \frac{12 \times 100 \times 11}{2 \times 13} \text{ pJ} = 507.69 \text{ pJ}$$

25. (d)

$$r_{t=5} = \text{area}$$

$$= \left(\frac{1}{2} \times 2 \times 2 + 2 \times 2 + 3 \times 1 \right) \text{ m}$$

$$= (2 + 4 + 3) \text{ m}$$

$$= 9 \text{ m}$$

$$26. (a) = \frac{\tau}{I}$$

$$= \frac{5M_0 g\ell - 4M_0 g\ell}{5M_0 \ell^2 + 2M_0 4\ell^2}$$

$$= \frac{M_0 g\ell}{13M_0 \ell^2}$$

$$= \frac{g}{13\ell}$$

$$27. (a) 2|\vec{P} + \vec{Q}| = |\vec{P} + 2\vec{Q}|$$

$$\Rightarrow 13 + 12\cos\theta = 10 + 6\cos\theta$$

$$\cos = -\frac{1}{2}$$

$$\theta = 120^\circ.$$

28. (b) (BONUS)

$$A\rho g x = F_{\text{restoring}}$$

$$\pi r^2 \rho g x = n\omega^2 x$$

$$\therefore \omega = \sqrt{\frac{\pi r^2 \rho g}{\rho v}}$$

$$= r \sqrt{\frac{\pi g}{v}} = 2.5 \times 10^{-2} \sqrt{\frac{3.14 \times 10}{310 \times 10^{-6}}}$$

$$= 2.5 \times 10^{-2} \times 10^2 \sqrt{10}$$

$$\omega = 2.5 \times \sqrt{10}$$

$$\therefore f = \frac{2.5 \times \sqrt{10}}{2\pi} = 1.25$$

29. (C)

$$|v_4| = |a_4|$$

$$\Rightarrow (w\sqrt{A^2 - x^2})_4 = (w^2x)_4$$

$$\Rightarrow w\sqrt{25 - 16} = w^2 \times 4$$

$$\Rightarrow w = \frac{3}{4}$$

$$T = \frac{2\pi}{W} = 2\pi \frac{4}{3} = \frac{8\pi}{3}$$

30. (c)

$$E = \frac{1}{2} MV_m^2$$

$$= \frac{1}{2} \times 2 \times \left(\frac{3P}{\rho}\right)$$

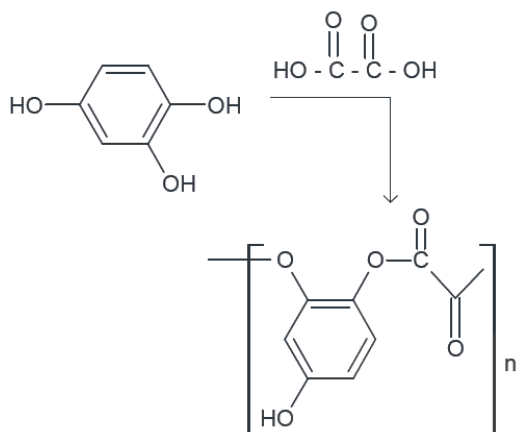
$$= \frac{3 \times 4 \times 10^4}{8} = 1.5 \times 10^4 \text{ J}$$

31. (c) $E = -13.6 \frac{n^2}{z^2} \text{ eV}$

$$E_{\text{He}}^+ = -13.6 \frac{4}{9}$$

32. (a)

33. (c)

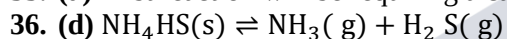


34. (c) $0.1 = \frac{n_{C_{12}H_{22}O_{11}}}{2}$

$n_{C_{12}H_{22}O_{11}} = 0.2$

$Wt_{C_{12}H_{22}O_{11}} = 0.2 \times 342 = 68.4$

35. (a) First reaction will be requiring a catalyst among halogens oxidizing power decrease down the group.



5.1 g

0.1 g mol – 0.03 0.03 mol 0.03 mol

$V = 3 \text{ L}, T = 327^\circ\text{C} \quad \frac{0.98}{2} \quad \frac{0.98}{2}$

$K_P = P_{NH_3} P_{H_2S} P_V = nRT$

$K_P = \frac{0.98}{2} \times \frac{0.98}{2} P \times 3 = 0.06 \times 0.0821 \times 600$

$K_P = 0.243$

$P = \frac{0.06 \times 0.0821 \times 200}{3}$

$P = 0.98$

37. (b) CH_4 is not present in stratosphere.

38. (d) $E = E^\circ - 0.06 \log \frac{[H^+][Cl^-]}{[H_2]^{1/2}}$

$0.92 = E^\circ - 0.06 \log \frac{10^{-6} \cdot 10^{-6}}{1^{1/2}}$

$0.92 = E^\circ - 0.06 \log 10^{-12}$

$E^\circ + 0.06 \times 12$

$E^\circ = 0.92 - 0.06 \times 12$

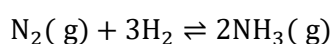
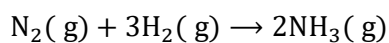
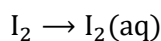
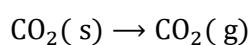
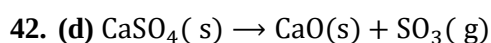
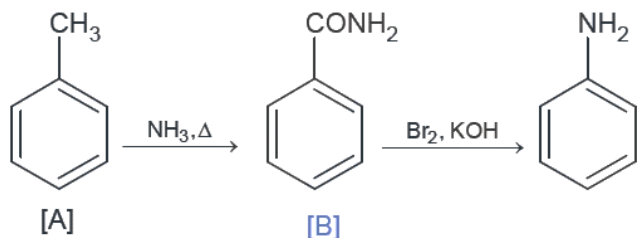
$E^\circ = 0.92 - 0.72$

$$E_{\text{Ag}}^{\circ}/\text{AgCl} = 0.20$$

39. (b) The electron configuration is $[Xe]4f^{14}5d^16s^2$

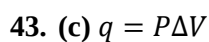
40. (a)

41. (a)



$$\Delta S = 2\Delta S_{\text{NH}_3} - [\Delta S_{\text{N}_2} + 3\Delta S_{\text{H}_2}]$$

There is decrease in number of moles of NH_3 entropy is decreasing.

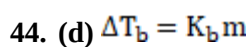


$q = 16$

$$C_P = 24$$

$$C_P = \frac{q_P}{\Delta T}$$

$$\Delta T = \frac{16}{24} \text{ K} = \frac{2}{3} \text{ K}$$



$$\Delta T_b = K_b \times 1 \Rightarrow 2 = K_b$$

$$\Delta T_f = K_f m$$

$$2 = 2 K_f = K_f$$

$$\frac{K_f}{K_b} = \frac{1}{2}$$

45. (d) NaBH_4 reduces both carbonyl group and imine.

46. (d)

47. (d)

$$\frac{1}{2} \frac{d[A]}{dt} = -K_1 [A]_2 + K_{-1} [A]^2$$

$$\frac{d[A]}{dt} = 2 K_1 [A_2] - 2 K_{-1} [A]^2$$

48. (b) Barfoed test is used to detect monosaccharides.

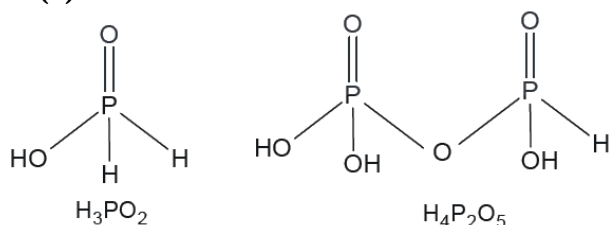
49. (c) Co^{2+} high spin $t_{2g}^5 e_g^2$ 3 ' unpaired electrons

Co^{2+} low spin $t_{2g}^6 e_g^1$ 1 ' unpaired electron

Difference is $3 - 1 = 2$

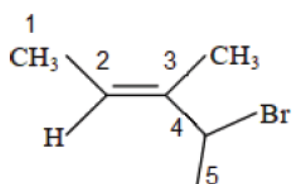
50. (d)

51. (b)



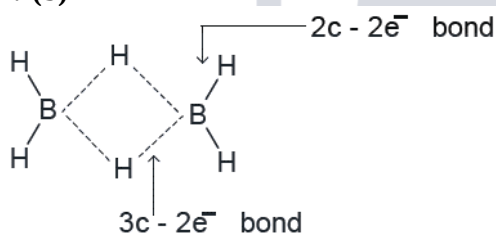
52. (b) Iodoform reaction can be used for this transformation.

53. (d)



4-Bromo-3-methyl pentane

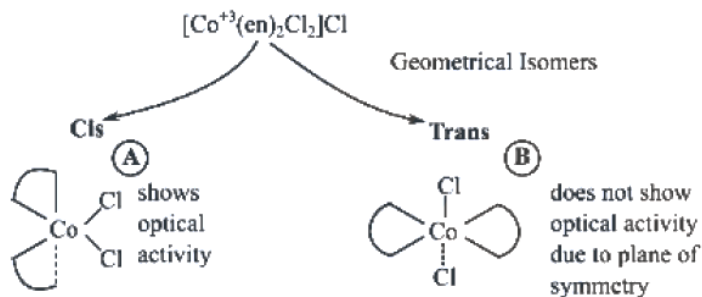
54. (b)



55. (a) $5\text{C}_2\text{O}_4^{2-} + 2\text{KMnO}_4 + \text{H}^+ \rightarrow 10\text{CO}_2 + 2\text{Mn}^{2+} + \text{H}_2\text{O}$

After balancing 10 gain or loss of electron.

56. (a) Cobalt (III) chloride on reaction with ethylenediamine in ratio 1:22 isomeric products complexes *A* and *B*



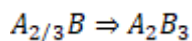
57. (a) $[\text{Au}(\text{CN})_2]^-$ and $[\text{Ag}(\text{CN})_2]^-$ both are soluble complexes.

58. (d) Formula of the compounds

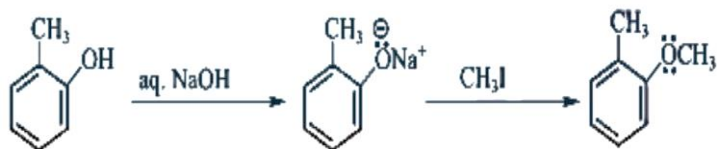
No. of octahedral voids are equal to the number of atoms forming lattice

A occupy octahedral void i.e. $\frac{2}{3}$ of them

B forms crystal lattice



59. (b)



60. (d) 'EAS' first ring is activated and second is deactivated NO_2^+ attack at para position of activated ring.

61. (d) $\alpha^2 + \beta^2 = (\alpha + \beta)^2 - 2\alpha\beta$

$$= (3 - \lambda)^2 - 2(2 - \lambda)$$

$$= \lambda^2 + 9 - 6\lambda - 4 + 2\lambda$$

$$= \lambda^2 - 4\lambda + 5$$

$\therefore$ For least value $\lambda = 2$

62. (a) Using formula $\frac{\sin 2^n A}{2^n \sin A} = \cos A \cos 2A \cos 2^2 A \dots \dots \dots \cos 2^{n-1} A$

63. (a) $x^2 dx + 2xy dy - y^2 dx = 0$

$$x^2 dx = y^2 dx - 2xy dy$$

$$(x^2 - y^2) dx + 2xy dy = 0$$

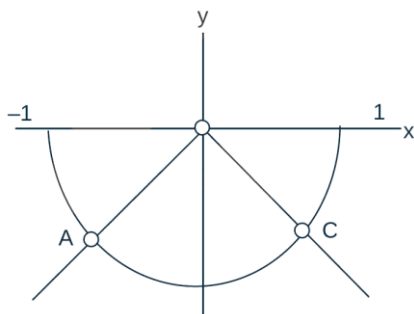
$$dx = - \left(\frac{x \cdot 2y dy - y^2 dx}{x^2} \right)$$

Integrals

$$x = -\frac{y^2}{x} + c$$

$$x^2 + y^2 = cx$$

64. (c) A, B, C are sharp edges



65. (a) $x^2 \left(\sqrt{x} + \frac{\lambda}{x^2} \right)^{10}$

Consider constant term

$${}^{10}C_r (\sqrt{x})^{10-r} \left(\frac{\lambda}{x^2}\right)^r$$

$$\frac{10-r}{2} - 2r = 0$$

$$10 - 5r = 0$$

$$r = 2$$

$$\Rightarrow {}^{10}C_2 \times \lambda^2 = 720 \Rightarrow \lambda = 4$$

66. (b) $y = xe^{x^2}$

(1, e) lies on this

Now $\frac{dy}{dx} = x^{x^2} \cdot 2x + e^{x^2} \cdot 1$

Put $x = 1$

$$m = 2e + e = 3e$$

Equation of tangent at (1, e)

$$y - e = 3e(x - 1)$$

$$y - e = 3ex - 3e$$

$$y = 3ex - 2e$$

$\left(\frac{4}{3}, 2e\right)$ satisfies it

67. (a) $\left. \begin{aligned} f(g(1)) &= 1 \\ f(g(2)) &= 1 \end{aligned} \right\}$ Many one

$$f(g(2k)) = k$$

$$f(g(2k+1)) = k+1$$

$\therefore$ Onto

68. (b)

$$\begin{vmatrix} \sin 3\theta & -1 & 1 \\ \cos 2\theta & 4 & 3 \\ 2 & 7 & 7 \end{vmatrix} = 0$$

$$7\sin 3\theta + 14\cos 2\theta - 14 = 0$$

$$\sin 3\theta + 2\cos 2\theta - 2 = 0, \sin \theta = \frac{1}{2}$$

69. (a) $\vec{\alpha} = (\lambda - 2)\vec{a} + \vec{b}$

$$\vec{\beta} = (4\lambda - 2)\vec{a} + 3\vec{b}$$

$\vec{\alpha}$ and $\vec{\beta}$ are collinear

$$\begin{vmatrix} \lambda - 2 & 1 \\ 4\lambda - 2 & 3 \end{vmatrix} = 0$$

$$3\lambda - 6 - 4\lambda + 2 = 0$$

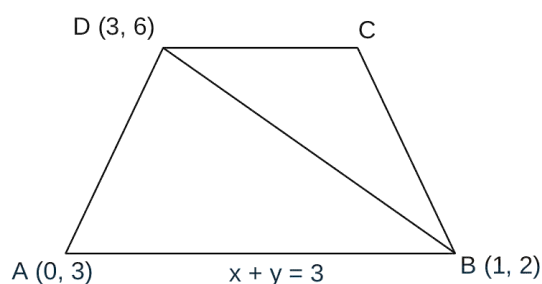
$$-\lambda - 4 = 0$$

$$\lambda = -4$$

70. (d)

Intersection point is $A(0,3)$ $M = (4,6)$

$B \Rightarrow (1,2), D \rightarrow (3,6)$



71. (a) Differentiability we get $f(x) = 2x - x^2 f(x)$

$$f(x) = \frac{2x}{1+x^2} \Rightarrow f''(x) = 2 \frac{(1-x^2)}{(1+x^2)^2}$$

$$f'\left(\frac{1}{2}\right) = \frac{24}{25}$$

72. (a)

$$\left(\frac{\sqrt{3}}{2} + \frac{i}{2}\right)^5 = \left(e^{i\frac{\pi}{6}}\right)^5 = e^{i5\pi/6}$$

$$\therefore z = 2\cos\frac{5\pi}{6} = 2\left(-\frac{\sqrt{3}}{2}\right) = -\sqrt{3}$$

73. (c)

$$1 - \left(\frac{2}{3}\right)^n > \frac{5}{6}$$

$$\left(\frac{2}{3}\right)^n < \frac{1}{6}$$

$$\Rightarrow n = 5$$

74. (b)

$$\int x^5 \cdot e^{-4x^3} dx$$

$$= \int x^2 \cdot x^3 e^{-4x^3} dx$$

$$\begin{aligned}
 -4x^3 &= t \\
 -12x^2 dx &= dt \\
 &= \frac{-1}{12} \int -\frac{t}{4} e^t dt \\
 &= \frac{1}{48} \int t e^t dt \\
 &= \frac{1}{48} t e^t - 1 \cdot e^t + c \\
 &= \frac{1}{48} e^{-4x^3} \cdot (-4x^3) - e^{-4x^3} + c
 \end{aligned}$$

75. (d) $r = \sqrt{25 + 36 - c} = \sqrt{36}$
 $c = 25$

76. (d) **P** is True

Q is False

R is True

$\therefore \sim Q$ is True

$\sim Q \wedge R$ is True

$\therefore P \vee (\sim Q \wedge R)$ is True.

77. (d) $x = \sqrt{2}y - 4\sqrt{2}$

$x^2 = 4y$

Solving we get point of intersection

$A(-2\sqrt{2}, 2), B(4\sqrt{2}, 8)$

$\therefore AB = \sqrt{(6\sqrt{2})^2 + 6^2} = 6\sqrt{3}$

78. (a) $\text{Det } A = b^2 + 3$

$\frac{\det A}{b} = b + \frac{3}{b}$

$\therefore \text{Least value} = 2\sqrt{3}$

79. (b) $\frac{y^2}{1+r} - \frac{x^2}{1-r} = 1$

$r > 1 \Rightarrow \text{ellipse}$

$e = \sqrt{1 - \left(\frac{r-1}{r+1}\right)} = \sqrt{\frac{2}{r+1}}$

80. (b)

$$\begin{aligned} & \sum_{r=1}^{25} \frac{50}{r! 50-r} \times \frac{50-r}{25-r} \\ &= \sum_{r=1}^{25} \frac{50}{r[25-r]25} \\ &= \frac{50}{25} \sum_{r=1}^{25} \frac{1}{r[25-r]} \\ &= \frac{50}{25} \sum_{r=1}^{25} {}^{25}C_r = {}^{50}C_{25} (2^{25} - 1) \end{aligned}$$

81. (d) $A(-3,3,4), B(3,7,6)$

Mid point $\Rightarrow (0,2,5)$

$$\vec{n} = \overrightarrow{AB} = 6\hat{i} + 10\hat{j} + 2\hat{k}$$

Equation of plane $\vec{r} \cdot \vec{n} = \vec{a} \cdot \vec{n}$

$$\vec{r} \cdot (6\hat{i} + 10\hat{j} + 2\hat{k}) = (0\hat{i} + 2\hat{j} + 5\hat{k}) \cdot (6\hat{i} + 10\hat{j} + 2\hat{k})$$

$$3x + 5y + z = 15$$

$(4,1,-2)$ satisfies it

82. (a) $\cot[\sum_{n=1}^{19} \cot^{-1}(1 + \sum_{p=1}^n 2p)]$

$$= \cot[\sum_{n=1}^{19} \cot^{-1}(1 + n^2 + n)]$$

$$= \cot \left[\sum_{n=1}^{19} \tan^{-1} \left(\frac{1}{1 + n^2 + n} \right) \right]$$

$$= \cot[\sum_{n=1}^{19} \tan^{-1}(n+1) - \tan^{-1} 1]$$

$$= \cot[\tan^{-1} 20 - \tan^{-1} 1]$$

$$= \cot \left(\tan^{-1} \frac{19}{21} \right)$$

$$\Rightarrow \frac{21}{19}$$

83. (d) $\sum x = 50$

$$(3)^2 = \frac{1}{5} \left(\sum x^2 - \frac{(\sum x)^2}{5} \right)$$

$$9 = \frac{1}{5} \left(\sum x^2 - \frac{2500}{5} \right)$$

$$\therefore \sum x^2 = 545$$

$$\text{New variable} = \frac{1}{6} \left(3045 - \frac{0}{6} \right) = 507.5$$

84. (b) $f'(x) = 7 - \frac{3}{4} \cdot \frac{f(x)}{x}, x > 0$

$$\therefore f'(x) + \frac{3}{4x}f(x) = 7$$

(Linear)

$$f(x) \cdot e^{\int \frac{3}{4x} dx} = \int 7 \cdot e^{\int \frac{3}{4x} dx} + c$$

$$f(x) \cdot x^{3/4} = \int 7 \cdot x^{3/4} + c$$

$$= 7 \frac{x^{7/4}}{\frac{7}{4}} + c$$

$$\therefore f(x) = 4x + cx^{-3/4}$$

$$\therefore f\left(\frac{1}{x}\right) = \frac{4}{x} + cx^{3/4}$$

$$\therefore \lim_{x \rightarrow 0^+} xf\left(\frac{1}{x}\right) = \lim_{x \rightarrow 0^+} 4 + cx^{7/4} = 4$$

(h, k)

85. (b)

$$\frac{k-3}{h-4} = 0 \quad k = 3$$

$$\frac{k}{h} = -\frac{4-0}{3-2} \quad -4h = k$$

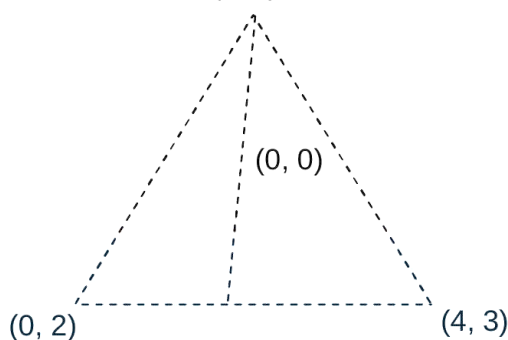
$$h = \frac{-3}{4}$$

(0, 2)

86. (c)

$$\int_{-\pi/2}^{\pi/2} \frac{dx}{[x] + [\sin x] + 4}$$

(h, k)



$$= \int_{-\pi/2}^0 \frac{dx}{[x] + -1 + 4} + \int_0^{\pi/2} \frac{dx}{[x] + 4}$$

$$= \int_{-\pi/2}^{-1} \frac{dx}{-2 - 1 + 4} + \int_{-1}^0 \frac{dx}{-1 - 1 + 4} + \int_0^1 \frac{dx}{4} + \int_1^{\pi/2} \frac{dx}{1 + 4}$$

$$= -1 + \frac{\pi}{2} + 2 + \frac{1}{4} + \frac{1}{5} \left(\frac{\pi}{2} - 1 \right)$$

$$= 3\frac{\pi}{5} - \frac{9}{20}$$

87. (c) Put $(2\lambda + 4, 2\lambda + 5, \lambda + 3)$ in $x + y + z = 2$

$$2\lambda + 4 + 2\lambda + 5 + \lambda + 3 = 2$$

$$5\lambda = -10 \quad \lambda = -2$$

P(0,1,1)

88. (b)

89. (a) $a = \sqrt{3} + 1$

$$b = \sqrt{3} - 1$$

$$\frac{\sin A}{\sin B} = \frac{\sqrt{3} + 1}{\sqrt{3} - 1} = \frac{3 + 1 + 2\sqrt{3}}{2} = 2 + \sqrt{3}$$

$$\frac{\sin A}{\sin(120 - A)} = \sqrt{3} + 2$$

$$\frac{\sin A}{\sin 12\cos A - \cos 12\sin A} = \sqrt{3} + 2$$

$$\frac{1}{\frac{\sqrt{3}}{2}\cot A + \frac{1}{2}} = \sqrt{3} + 2$$

$$\frac{\sqrt{3}\cot A + 1}{2} = \frac{1}{\sqrt{3} + 2} = \frac{\sqrt{3} - 2}{-1}$$

$$\frac{\sqrt{3}\cot A + 1}{2} = -\sqrt{3} + 2$$

$$\sqrt{3}\cot A = 4 - 2\sqrt{3} - 1$$

$$\sqrt{3}\cot A = 3 - 2\sqrt{3}$$

$$\cot A = \sqrt{3} - 2$$

$$-\cot A = 2 - \sqrt{3} = \tan 15^\circ$$

$$\therefore A = 105^\circ$$

$$\therefore B = 15^\circ$$

90. (c) $y = x^{3/2} - 2 \quad \frac{dy}{dx} = \frac{3}{2}\sqrt{x}$

$$\text{Slope of normal} = -\frac{2}{3\sqrt{x}}$$

$$\text{Let point is } (x_1, x_1^{3/2} - 2)$$

$$\therefore \text{Normal } y - \left(x_1^{3/2} - 2\right) = \frac{-2}{3\sqrt{x_1}}(x - x_1)$$

Now put (1,7) and solve it.

$$\Rightarrow x_1 = \frac{1}{3}$$

$$\therefore P \Rightarrow \left(\frac{1}{3}, 7 + \frac{1}{3\sqrt{3}}\right), A \Rightarrow (1,7)$$

$$\therefore AD = \frac{1}{6}\sqrt{\frac{7}{3}}$$

